

Gold-Catalyzed Stereocontrolled Oxacyclization/[4+2]-Cycloaddition Cascade of Ketone-Allene Substrates

Tse-Min Teng and Rai-Shung Liu*

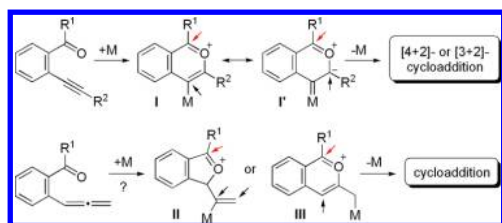
Department of Chemistry, National Tsing Hua University, Hsinchu, Taiwan, ROC

Received May 20, 2010; E-mail: rslu@mx.nthu.edu.tw

Abstract: We report the first success on the Au-catalyzed tandem oxacyclization/[4+2]-cycloaddition cascade using ketone-allene substrates to give highly substituted oxacyclics with excellent stereocontrol. In contrast to oxo-alkyne substrates, the resulting cycloadducts are isolable and efficiently produced from a reasonable scope of enol ethers.

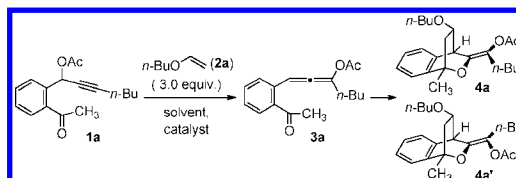
Metal-catalyzed cycloaddition/annulation reactions are important tools to access complex molecular frameworks.¹ The Au- and Pt-catalyzed activation of alkynes enables the generation of unusual intermediates to react with dipolarophiles in a cycloaddition fashion.² Such reactions have attained considerable success only on oxo-alkyne substrates.^{1a,b} In the presence of Au or Pt catalysts, 2-oxo-1-alkynylbenzenes form metal-containing benzopyrylium (**I**) or carbonyl ylides (**I'**) that react with dipolarophiles to give hypothetical [4+2]- or [3+2]-cycloaddition intermediates.^{3,4} Generation of 1,n-dipole species remains unexplored for oxo-allene substrates. This approach is mechanistically appealing because of the uncertain workability of oxonium-vinylmetal (**II**) or benzopyrylium (**III**) as 1,n-dipoles ($n = 4, 5$). Species **II** is kinetically favored by the Pt or Au- π -allene bonding,⁵ but its participation is absent in this work.

Scheme 1



We prepared alkynyl acetate **1a** readily from 2'-bromoacetophenone. This substrate works as a precursor to generate ketone-allene **3a** via a catalytic 1,3-acyloxy shift.⁶ As shown in Table 1, treatment of compound **1a** with *n*-butyl vinyl ether (**2a**, 3 equiv) with PtCl₂/CO or AuCl₃ in hot dichloroethane (50–80 °C) gave an exclusive recovery of unreacted **1a**. The use of PPh₃AuCl/AgSbF₆ (3 mol %) in dichloromethane (DCM) produced ketone-allene **3a** in 83% yield, but with no tractable amount of cycloadducts **4a/4a'**. We were pleased to discover that ClAuP(*t*-Bu)₂(*o*-biphenyl)/AgSbF₆ (3 mol %) enabled the desired cyclization/cycloaddition cascade to give **4a** (29%) and **4a'** (56%) as a diastereomeric mixture, separable on a silica column. The stereoselectivity is greatly enhanced with ClAuP(*t*-Bu)₂(*o*-biphenyl)/AgNTf₂, giving only **4a'** in 79% yield (entry 5); within a brief time (1 h), we obtained ketone-allene **3a** in 40% yield in addition to the desired **4a'** (46%, entry 6). The intermediacy of ketone-allene **3a** was confirmed by a complete

Table 1. Screening of Catalytic Activity over Various Metal Catalysts

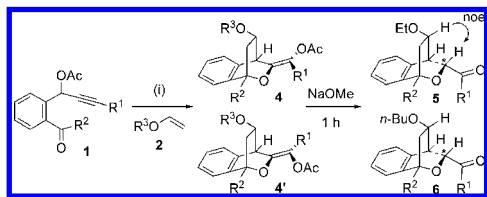


entry	substrate ^a	catalyst (mol %)	condition	product (yield) ^b
1	1a	AuCl ₃ (5)	DCE (80 °C, 24 h)	1a (85%)
2	1a	PtCl ₂ /CO (5)	DCE (50 °C, 20 h)	1a (46%) ^c
3	1a	Ph ₃ PAuCl (3)/AgSbF ₆ (3)	DCM (30 °C, 8 h)	3a (83%)
4	1a	AuCIL (3)/AgSbF ₆ (3)	DCM (30 °C, 2 h)	4a (29%), 4a' (56%)
5	1a	AuCIL (3)/AgNTf ₂ (3)	DCM (30 °C, 2.5 h)	4a' (trace), 4a' (79%)
6	1a	AuCIL (3)/AgNTf ₂ (3)	DCM (30 °C, 1 h)	3a (40%), 4a' (46%)
7	3a	AuCIL (3)/AgNTf ₂ (3)	DCM (30 °C, 2 h)	4a (trace), 4a' (86%)
8	1a	AuCIL (5)	DCM (30 °C, 20 h)	1a (94%)
9	1a	AgNTf ₂ (5)	DCM (30 °C, 20 h)	1a (62%), 3a (17%)
10	1a	IPrAuCl (3)/AgNTf ₂ (3)	DCM (30 °C, 2 h)	4a (48%), 4a' (22%)

^a L = P(*t*-Bu)₂(*o*-biphenyl), [substrate] = 0.1 M, DCM = dichloromethane, DCE = 1,2-dichloroethane. ^b Isolated yields were reported after purification from a silica gel column. ^c Decomposition of starting **1a** was observed in entry 2.

conversion to **4a'** with the same gold catalyst (entry 7). Control experiments indicate that ClAuP(*t*-Bu)₂(*o*-biphenyl) and AgNTf₂ alone were catalytically inactive (entries 8–9). An altered chemoselectivity was observed for IPrAuCl/AgNTf₂ (IPr = 1,3-bis(diisopropyl-phenyl)imidazol-2-ylidene) that preferably gave **4a** as the major product (entry 10). Characterization of the structure of product **4a'** relies on an X-ray diffraction study of its analogue **4b'** (Table 2, entry 1).

We prepared ketone substrates **1b–1e** bearing altered R¹ and R² substituents to examine the scope of this catalysis, as depicted in Table 2. Herein, the aldehyde substrates are not studied due to the intrinsic instability of 2-allenyl benzaldehyde intermediates.⁷ This reaction works well with both ethyl- and *n*-butyl vinyl ether (**2a–2b**), and it is also efficient with alterations of the R¹ (Me, *n*-Bu, *i*-Bu) and R² alkyls (Me, *n*-Pr) of substrates **1a–1e**. In entries 2, 4, and 5, due to the poor diastereoselectivity or chromatographic inseparable property, the cycloadducts **4b/4b'** and **4c/4c'** were subject to deacylation with NaOMe in MeOH to give ketone derivatives **5b**, **6b**, and **6c** with high dr values (8.5–10.0:1). For

Table 2. Au(I)-Catalyzed Oxacyclization/Cycloaddition Cascade with Vinyl Ethers

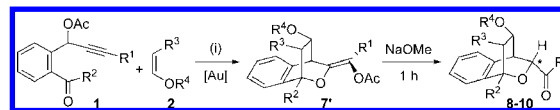
entry	substrates ^a	enol ether	time (step i)	products (yield) ^b
1	R ¹ = R ² = Me (1b)	R ³ = Et (2b)	1.5 h	4b (46%), 4b' (42%)
2	1b	2b	1.5 h	5b (85%, dr = 9:1) ^c
3	R ¹ = <i>n</i> -Bu, R ² = Me (1a)	R ³ = <i>n</i> -Bu (2a)	2 h	4a' (79%, dr > 20:1)
4	R ¹ = Me, R ² = Me (1b)	2a	2 h	6b (83%, dr = 10:1) ^c
5	R ¹ = Me, R ² = <i>n</i> -Pr (1c)	2a	2 h	6c (76%, dr = 8.5:1) ^c
6	R ¹ = <i>n</i> -Bu, R ² = <i>n</i> -Pr (1d)	2a	2 h	4d' (69%, dr > 20:1)
7	R ¹ = <i>i</i> -Bu, R ² = Me (1e)	2a	3 h	4e' (61%, dr > 20:1) ^d

^a ClAuP(*t*-Bu)₂(*o*-biphenyl)/AgNTf₂ (3 mol %), [substrate] = 0.1 M, DCM, enol ethers (3 equiv). ^b Isolated yields were reported after purification from a silica gel column. ^c The configuration at the *C carbon is responsible for the occurrence of two diastereomers. ^d Messy mixture was obtained for portion of **1e**.

substrates **1a** and **1d–1e** bearing a bulky R¹ (R¹ = *n*-Bu or *i*-Bu), the same reactions gave compounds **4a'** and **4d'–4e'** with excellent dr values (>20:1), due to a large steric interaction of the other isomers **4**. The structure of representative compound **4b'** is determined by an X-ray diffraction study.⁸

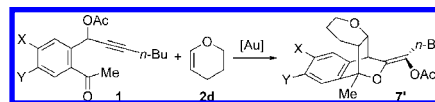
The scope of this new synthetic method is substantially expanded with its compatibility with substituted enol ethers including 1-ethoxypro-1-ene (**2c**, *Z/E* = 1.8, 3 equiv) and cyclic enol ethers **2d–2e**, as illustrated in Table 3. These enol ethers proceeded with excellent diastereoselectivity that we did not obtain any epimers bearing mutual *trans*-R³ and OR⁴ substituents. For starting ketones **1a–1c** and 1-ethoxypro-1-ene (entries 1–3), their cycloadducts **8a–8c** were obtained as one single diastereomer with its structure carefully determined by ¹H-NOE, suggesting that *cis*-enol ether **2c** is more active than its *trans* isomer. Indeed, the reaction of **2c** (*Z/E* = 0.5, 3 equiv) with ketone **1a** gave cycloadduct **8a** in diminished yield (31%) together with ketone-allene **3a** (61%). This new tandem cascade also works with 3,4-dihydro-2*H*-pyran (**2d**) that reacts with ketones **1a–1f** smoothly (entries 5–10), giving satisfactory yields (58–82%) of the expected cycloadducts **7a'**, **7d'–7f'** or the deacylation products **9b–9c**. 2,3-Dihydrofuran is also applicable to this catalysis, and it reacts with ketone **1b** to deliver compound **10b** in 53% yield (dr = 6.5:1).

We also prepared new substrates **1g–1l** to examine the effects of their phenyl substituents; their reactions with 3,4-dihydro-2*H*-pyran are described in Table 4. Good yields (72–87%) were obtained for cycloadducts **7g'–7h'** and **7k'–7l'** bearing fluoro and methoxy substituents because these π -donor groups stabilize proposed benzopyriliums **III** (*vide infra*). Hypothetic [4+2]-cycloadditions of free benzopyriliums^{3f–h} and their metal-containing analogues³ are restricted to unsubstituted benzenes and methoxy derivatives. The workability with substrates **1i** and **1j** bearing electron-deficient benzenes highlights the high reactivity of our new benzopyrilium (**III**).

Table 3. Au(I)-Catalyzed Oxacyclization/Cycloaddition Cascade with Substituted Enol Ethers

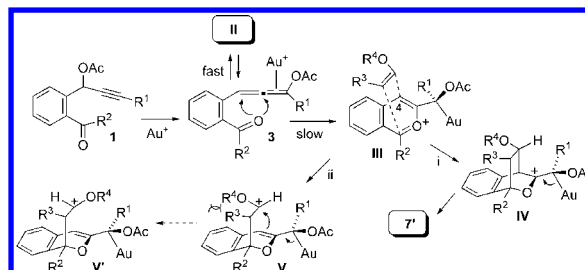
entry	substrates ^a	alkene	t (step i)	products ^b
1	R ¹ = <i>n</i> -Bu, R ² = Me (1a)	R ³ = Me, R ⁴ = Et (2c , <i>Z/E</i> = 1.8)	4 h	8a (76%, dr > 20:1) ^c
2	R ¹ = R ² = Me (1b)	2c	4 h	8b (82%, dr = 7.3:1) ^c
3	R ¹ = Me, R ² = <i>n</i> -Pr (1c)	2c	4 h	8c (81%, dr > 15:1) ^c
4	R ¹ = <i>n</i> -Bu, R ² = Me (1a)	2c (<i>Z/E</i> = 0.5)	4 h	8a (31%, dr > 20:1) ^{c,d}
5	R ¹ = <i>n</i> -Bu, R ² = Me (1a)	R ³ , R ⁴ = -(CH ₂) ₃ - (2d)	1 h	7a' (82%, dr > 20:1)
6	R ¹ = R ² = Me (1b)	2d	2 h	9b (72%, dr = 10:1) ^c
7	R ¹ = Me, R ² = <i>n</i> -Pr (1c)	2d	1 h	9c (64%, dr = 7.5:1) ^c
8	R ¹ = <i>n</i> -Bu, R ² = <i>n</i> -Pr (1d)	2d	2 h	7d' (75%, dr > 20:1)
9	R ¹ = <i>i</i> -Bu, R ² = Me (1e)	2d	3 h	7e' (63%, dr > 20:1) ^e
10	R ¹ = Ph, R ² = Me (1f)	2d	5 h	7f' (58%, dr > 20:1) ^e
11	R ¹ = R ² = Me (1b)	R ³ , R ⁴ = -(CH ₂) ₂ - (2e)	2 h	10b (53%, dr = 6.5:1) ^{c,e}

^a ClAuP(*t*-Bu)₂(*o*-biphenyl)/AgNTf₂ (3 mol %), [substrate] = 0.1 M, DCM, enol ethers (3 equiv). ^b Isolated yields were given after purification from a silica gel column. ^c The configuration at the *C carbon is responsible for the occurrence of two diastereomers. ^d Ketone-allene **3a** was obtained in 61%. ^e Messy mixture was obtained for portion of **1b**, **1e**, or **1f**.

Table 4. Au(I)-Catalyzed Oxacyclization/Cycloaddition Cascade of Substituted Aromatic Substrates

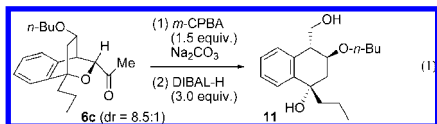
entry	substrates ^a	t [h]	products (yield) ^b
1	X = H, Y = F (1g)	6	7g' (78%, dr > 20:1)
2	X = F, Y = H (1h)	8	7h' (72%, dr > 20:1)
3	X = H, Y = Cl (1i)	6	7i' (42%, dr > 20:1)
4	X = Cl, Y = H (1j)	6	7j' (47%, dr > 20:1)
5	X = H, Y = OMe (1k)	2	7k' (87%, dr > 20:1)
6	X = OMe, Y = H (1l)	2	7l' (85%, dr > 20:1)

^a ClAuP(*t*-Bu)₂(*o*-biphenyl)/AgNTf₂ (3 mol %), [substrate] = 0.1 M, DCM, enol ethers (3 equiv). ^b Isolated yields were given after purification from a silica gel column.

Scheme 2

Structural analysis of resulting cycloadducts asserts the intermediacy of benzopyrilium (**III**), although gold- π -allene species **3** has cationic character located mainly at the C(1)- and C(3)-carbons.⁵ We hypothesize a fast equilibrium between ketone-allene (**3**) and intermediate (**II**), but the cycloadducts expected from species (**II**) fail to proceed. Here, we propose a concerted mechanism (i) for

the [4+2]-cycloadditions of the new benzopyrilium (**III**). As shown in Scheme 2, *cis*-substituted enol ether approaches the pyrilium core of species **III** with its R³ and OR⁴ lying away from the bulky gold-containing substituent. We envisage that the allylic gold fragment of benzopyrilium **III** raises the HOMO energy level at the C(4)-carbon, facilitating this concerted process.^{2b,9} The stepwise pathway (ii) involving cationic intermediates **V** and **V'** is opposed by our observation that no epimers resulted from the conformer **V'** which is actually favored by steric interactions of the *cis*-R³ and OR⁴ substituents of species **V**. Species (**III**) is more useful than reported benzopyriliums³ in synthetic utility, because of its isolable and stereocontrolled [4+2]-cycloadducts.



Equation 1 shows the use of this catalysis for a stereoselective synthesis of a highly oxygenated molecule. A sequential treatment of compound **6c** with *m*-CPBA (1.5 equiv), followed by the DIBAL-H cleavage of resulting acetal, gave triol derivative **11** (65%) as a single diastereomer.

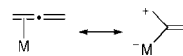
In summary, we report the first success on the Au-catalyzed tandem oxacyclization/[4+2]-cycloaddition cascade using ketone-allene substrates to give highly substituted oxacyclics with excellent stereocontrol. Control experiments reveal the involvement of benzopyrilium intermediates (**III**) that is active for [4+2]-cycloaddition reactions.⁹ In contrast to oxo-alkyne substrates,³ the resulting cycloadducts are isolable and efficiently produced from a reasonable scope of enol ethers. Efforts to realize the asymmetric version of this catalysis is under current investigation.

Acknowledgment. The authors wish to thank the National Science Council, Taiwan for supporting this work.

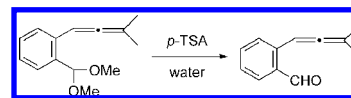
Supporting Information Available: Experimental procedures, characterization data of new compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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